

This article was downloaded by:

On: 22 January 2011

Access details: *Access Details: Free Access*

Publisher *Taylor & Francis*

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Asian Natural Products Research

Publication details, including instructions for authors and subscription information:

<http://www.informaworld.com/smpp/title~content=t713454007>

A new decalin derivative from red yeast rice

Yun-Tian Zhang^a; Ying Wang^{bc}; Xian-Tao Zhang^d; Dong-Ling Wu^a; Xiao-Qi Zhang^{bc}; Wen-Cai Ye^{abc}

^a Department of Phytochemistry, China Pharmaceutical University, Nanjing, China ^b Institute of Traditional Chinese Medicine and Natural Products, Jinan University, Guangzhou, China ^c Guangdong Province Key Laboratory of Pharmacodynamic Constituents of TCM and New Drugs Research, Jinan University, Guangzhou, China ^d Guangdong Research Institute of Traditional Chinese Medicine, Guangzhou, China

To cite this Article Zhang, Yun-Tian , Wang, Ying , Zhang, Xian-Tao , Wu, Dong-Ling , Zhang, Xiao-Qi and Ye, Wen-Cai(2009) 'A new decalin derivative from red yeast rice', Journal of Asian Natural Products Research, 11: 9, 792 – 795

To link to this Article: DOI: 10.1080/10286020903164269

URL: <http://dx.doi.org/10.1080/10286020903164269>

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: <http://www.informaworld.com/terms-and-conditions-of-access.pdf>

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

A new decalin derivative from red yeast rice

Yun-Tian Zhang^a, Ying Wang^{bc}, Xian-Tao Zhang^d, Dong-Ling Wu^a, Xiao-Qi Zhang^{bc} and Wen-Cai Ye^{abc*}

^aDepartment of Phytochemistry, China Pharmaceutical University, Nanjing 210009, China;

^bInstitute of Traditional Chinese Medicine and Natural Products, Jinan University, Guangzhou 510632, China; ^cGuangdong Province Key Laboratory of Pharmacodynamic Constituents of TCM and New Drugs Research, Jinan University, Guangzhou 510632, China; ^dGuangdong Research Institute of Traditional Chinese Medicine, Guangzhou 510632, China

(Received 7 April 2009; final version received 5 July 2009)

A new decalin derivative, monascusic acid A (**1**), together with a new natural product (**2**), was isolated from the ethanol extract of red yeast rice. Their structures were elucidated by spectroscopic methods.

Keywords: monascusic acid A; decalin; *Monascus purpureus*; red yeast rice

1. Introduction

Red yeast rice, also called as ‘red mold rice’ or ‘Hongqu’, has been used as food and traditional medicine for a long time in China [1–5]. Red yeast rice could be obtained by the fermentation of rice with fungi of the genus *Monascus* (Monascaeae). Recently, more and more attention has been paid to the secondary metabolic products of *Monascus* species especially after the discovery of monacolins, which are potent inhibitors of HMG-CoA reductase and can lower blood lipid levels in both animal models and humans [1,5,6]. In our search for biologically active and structurally unique compounds from traditional Chinese medicine, we carried out the chemical investigation on red yeast rice (fermented by *Monascus purpureus*), resulting in the isolation of a new decalin derivative (**1**) and a new natural product (**2**) (Figure 1). Their structures were determined based on spectroscopic data.

Herein, we report the structure elucidation of **1**.

2. Results and discussion

Compound **1** was obtained as colorless needles, mp 119–120°C. The molecular formula of **1** was determined to be C₁₅H₂₂O₂ by HR-ESI-MS at *m/z* 233.1545 [M–H][–]. In the UV spectrum of **1**, the absorption maxima at 231, 238, and 247 nm indicated that **1** possessed a conjugated double bond [7]. The IR spectrum revealed the presence of hydroxyl (3429 cm^{–1}) and carbonyl (1698 cm^{–1}) groups. The ¹H NMR spectrum of **1** displayed two methyl signals at δ_H 0.90 (3H, d, *J* = 7.0 Hz, H-9) and 0.99 (3H, d, *J* = 7.1 Hz, H-10), and three olefinic protons at δ_H 5.44 (1H, m, H-5), 5.72 (1H, dd, *J* = 9.6, 5.9 Hz, H-3), and 5.91 (1H, d, *J* = 9.6 Hz, H-4). The ¹³C NMR spectrum showed 15 carbon signals, including two methyl, four methylene, seven methine, and two quaternary

*Corresponding author. Email: chywc@yahoo.com.cn

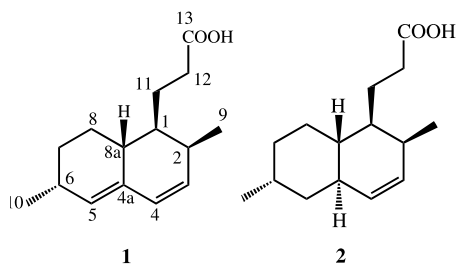


Figure 1. Chemical structures of **1** and **2**.

carbons. Comparison of the ^{13}C NMR spectral data of **1** with those of monacolin L reported in the literature [8,9] revealed that they were very similar except for the presence of a carbonyl carbon signal at δ_{C} 180.0 (C-13) in **1** instead of signals for the lactone ring. The combined analysis of 1D and 2D NMR spectral data of **1** suggested that compound **1** possessed a 3,4a(5)-diene decalin skeleton. The cross-peaks between H-11 (δ_{H} 1.45 and 2.04) and H-12 (δ_{H} 2.25 and 2.44) in the ^1H - ^1H COSY spectrum of **1**, and the HMBC correlations between H-12 (δ_{H} 2.25 and 2.44) and C-11 (δ_{C} 24.4) and C-13 (δ_{C} 180.0) indicated the presence of a propionic acid moiety. With the aid of ^1H - ^1H COSY, HSQC, and HMBC experiments, all the proton and carbon signals of **1** were assigned as shown in Table 1.

The HMBC correlations between H-1 (δ_{H} 1.45) and C-12 (δ_{C} 31.7) and between H-11 (δ_{H} 1.45 and 2.04) and C-2 (δ_{C} 31.4) indicated that the propionic acid moiety should be attached to the C-1 position of the decalin skeleton. Furthermore, in the HMBC spectrum, the correlations between H-9 (δ_{H} 0.90) and C-2 (δ_{C} 31.4) and C-3 (δ_{C} 132.8), as well as between H-10 (δ_{H} 0.99) and C-5 (δ_{C} 130.7) and C-7 (δ_{C} 29.3), indicated that the two methyl groups were located at the C-2 and C-6 positions of the decalin skeleton, respectively.

The relative stereochemistry of compound **1** could be determined on the basis of the ROESY correlations (Figure 2). The ROESY correlations between H-7 β and H-8 α , and between H-8 α and H-9, indicated the β -orientation for H-8 α and

Me-2. Similarly, the ROESY correlations between H-8 α and H-10, H-8 α and H-1, as well as between H-1 and H-2 suggested that Me-6, H-1, and H-2 should be α -oriented. Thus, the structure of **1** was determined as 3-(2,6-dimethyl-1,2,6,7,8,8a-hexahydronaphthalen-1-yl)-propanoic acid and named as monascusic acid A.

Compound **2** had been previously obtained from the cultures of *Aspergillus nidulans* mutant with controlled lovastatin biosynthesis gene [10]. In our present investigation, it was isolated and reported as a new natural product with full assignment of NMR spectral data for the first time.

3. Experimental

3.1 General experimental procedures

Melting points were determined on an XT-4 micro-melting point apparatus and were uncorrected. Optical rotation values were recorded on a JASCO P-1020 polarimeter with a 0.1 dm length cell at room temperature. UV spectra were measured on a JASCO V-550 UV/vis spectrophotometer with a 1 cm length cell. IR spectra (KBr) were recorded on a JASCO FT/IR-480 plus Fourier transform infrared spectrometer. HR-ESI-MS were obtained on an Agilent 6210 LC-MSD TOF mass spectrometer. NMR spectra were measured on a Bruker AV-400 spectrometer using tetramethylsilane as the internal standard.

Precoated silica gel GF₂₅₄ plates for TLC and silica gel (200–300 mesh) for column chromatography (CC) were obtained from Qingdao Haiyang Chemical Group Co. Ltd (Qingdao, China). All other chemical reagents were purchased from Shanghai Chemical Reagent Co. Ltd (Shanghai, China).

3.2 Preparation of red yeast rice

Red yeast rice was obtained by fermentation of moistened rice, with a strain

Table 1. ^1H and ^{13}C NMR spectral data of **1** and **2** (CDCl_3 , δ , J in Hz).^a

Position	1		2	
	δ_{C}	δ_{H}, J	δ_{C}	δ_{H}, J
1	41.5 (CH)	1.45	39.9 (CH)	1.01
2	31.4 (CH)	2.30	31.9 (CH)	2.24
3	132.8 (CH)	5.72 (dd, $J = 9.6, 5.9$)	132.4 (CH)	5.59 (ddd, $J = 9.8, 4.8, 2.7$)
4	128.4 (CH)	5.91 (d, $J = 9.6$)	131.6 (CH)	5.30 (br d, $J = 9.8$)
4a	136.4 (C)	—	37.3 (CH)	1.96
5	130.7 (CH)	5.44 (m)	38.9 (CH ₂)	1.47 (m, H α), 1.27 (td, $J = 13.0, 4.8, \text{H}\beta$)
6	28.7 (CH)	2.33	27.5 (CH)	1.98 (m)
7	29.3 (CH ₂)	1.56 (H α)	32.3 (CH ₂)	1.56
8	22.5 (CH ₂)	1.74 (dq, $J = 13.5, 3.0, \text{H}\beta$) 1.20 (br d, $J = 13.5, \text{H}\alpha$)	23.6 (CH ₂)	1.13 (qd, $J = 12.4, 4.1, \text{H}\alpha$) 1.56 (H β)
8a	34.9 (CH)	1.79 (m, H β)	41.0 (CH)	1.56
9	13.8 (CH ₃)	2.08	14.9 (CH ₃)	0.85 (d, $J = 7.0$)
10	21.2 (CH ₃)	0.90 (d, $J = 7.0$)	18.2 (CH ₃)	0.98 (d, $J = 7.2$)
11	24.4 (CH ₂)	0.99 (d, $J = 7.1$)	23.8 (CH ₂)	1.38 (m), 1.96
12	31.7 (CH ₂)	1.45, 2.04	32.0 (CH ₂)	2.24, 2.42 (ddd, $J = 15.7, 10.3, 5.2$)
13	180.0 (C)	2.25, 2.44 (ddd, $J = 15.4, 9.8, 4.9$)	180.3 (C)	—

Note: ^aOverlapped signals are reported without designating multiplicity.

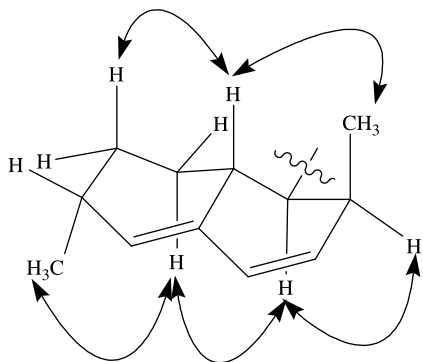


Figure 2. The key ROESY correlations of **1** of *M. purpureus* for 9 days at 25°C, at a pH range of 5–6.

3.3 Extraction and isolation

Dried and powdered red yeast rice (10 kg) was extracted three times with 95% ethanol at room temperature (3× 10 liters). The extracts were combined and concentrated under vacuum to yield the crude extract (800 g). The residue was suspended in 20% ethanol, and then successively partitioned with petroleum ether and ethyl acetate, respectively. After removing the solvent, the ethyl acetate extract (500 g) was separated by silica gel CC using gradient mixtures of *n*-hexane–ethyl acetate (100:0 → 0:100) as eluants to yield 15 fractions (1–15). Fraction 2 was then subjected to silica gel CC eluting with *n*-hexane–ethyl acetate (100:0 → 10:1) to afford **1** (60 mg) and a mixture. The mixture containing compound **2** was recrystallized with ethyl acetate to yield **2** (50 mg).

3.3.1 Compound 1

Colorless needles, mp 119–120°C; $[\alpha]_D^{20} + 220.2$ ($c = 0.14$, MeOH); UV (MeOH) λ_{\max} (log ϵ): 231 (4.30), 238 (4.41), 247 (4.20) nm; IR (KBr) ν_{\max} : 3429 (OH),

2964 and 2918 (CH), 1698 (C=O), 1446 (CH₂), 1371 (CH₃), 955, 860, 763, 615 cm⁻¹; ¹H and ¹³C NMR spectral data, see Table 1; HR-ESI-MS: m/z 233.1545 [M–H]⁻ (calcd for C₁₅H₂₁O₂, 233.1547).

3.3.2 Compound 2

Colorless needles, mp 131–132°C; $[\alpha]_D^{20} + 118.7$ ($c = 0.11$, MeOH); UV (MeOH) λ_{\max} (log ϵ): end absorption; IR (KBr) ν_{\max} : 3020, 2964, and 2908 (CH), 1697 (C=O), 1446 (CH₂), 1378 (CH₃), 956, 721 cm⁻¹; ¹H and ¹³C NMR spectral data, see Table 1; HR-ESI-MS: m/z 235.1700 [M–H]⁻ (calcd for C₁₅H₂₃O₂, 235.1704).

Acknowledgements

This research was supported by grants from the National Natural Science Foundation for Outstanding Young Scientists (No. 30625039) and Program for Changjiang Scholars (to Dr Ye).

References

- [1] P. Jůzlová, L. Martínková, and V. Křen, *J. Ind. Microbiol.* **16**, 163 (1996).
- [2] J. Ma, Y. Li, Q. Ye, J. Li, Y. Hua, D. Ju, D. Zhang, R. Cooper, and M. Chang, *J. Agric. Food Chem.* **48**, 5220 (2000).
- [3] T. Akihisa, S. Mafune, M. Ukiya, Y. Kimura, K. Yasukawa, T. Suzuki, H. Tokuda, N. Tanabe, and T. Fukuoka, *J. Nat. Prod.* **67**, 479 (2004).
- [4] D. Wild, G. Tóth, and H.U. Humpf, *J. Agric. Food Chem.* **51**, 5493 (2003).
- [5] C. Li, Y. Zhu, Y. Wang, J.S. Zhu, J. Chang, and D. Kritchevsky, *Nutr. Res.* **18**, 71 (1998).
- [6] A. Endo, *J. Antibiot.* **33**, 334 (1980).
- [7] A. Endo, *J. Antibiot.* **32**, 852 (1979).
- [8] A. Endo, K. Hasumi, and S. Negishi, *J. Antibiot.* **38**, 420 (1985).
- [9] A. Endo, K. Hasumi, T. Nakamura, M. Kunishima, and M. Masuda, *J. Antibiot.* **38**, 321 (1985).
- [10] J.L. Sorensen, K. Auclair, J. Kennedy, C.R. Hutchinson, and J.C. Vederas, *Org. Biomol. Chem.* **1**, 50 (2003).